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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/586,487	07/20/2006	Kousuke Chiba	2006_1165A	4198
	7590 10/15/200 [, LIND & PONACK, I	EXAMINER		
2033 K STREET N. W. SUITE 800 WASHINGTON, DC 20006-1021			BARRY, CHESTER T	
			ART UNIT	PAPER NUMBER
	•		1797	
		MAIL DATE	DELIVERY MODE	
			10/15/2008	PAPER

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Office Action Summary		Application No.	Applicant(s)			
		10/586,487	CHIBA, KOUSUKE			
		Examiner	Art Unit			
		CHESTER T. BARRY	1797			
Period fo	The MAILING DATE of this communication app or Reply	ears on the cover sheet with the c	orrespondence address			
WHIC - Exter after - If NO - Failu Any r	ORTENED STATUTORY PERIOD FOR REPLY CHEVER IS LONGER, FROM THE MAILING DATES as ions of time may be available under the provisions of 37 CFR 1.13 SIX (6) MONTHS from the mailing date of this communication. In period for reply is specified above, the maximum statutory period were to reply within the set or extended period for reply will, by statute, reply received by the Office later than three months after the mailing and patent term adjustment. See 37 CFR 1.704(b).	ATE OF THIS COMMUNICATION 36(a). In no event, however, may a reply be time will apply and will expire SIX (6) MONTHS from a cause the application to become ABANDONE	the mailing date of this communication. D (35 U.S.C. § 133).			
Status						
1)🛛	Responsive to communication(s) filed on 20 July 2006.					
2a) <u></u>	This action is FINAL . 2b)⊠ This	action is non-final.				
3)	Since this application is in condition for allowance except for formal matters, prosecution as to the merits is					
	closed in accordance with the practice under <i>Ex parte Quayle</i> , 1935 C.D. 11, 453 O.G. 213.					
Dispositi	on of Claims					
 4) Claim(s) 8-11 is/are pending in the application. 4a) Of the above claim(s) is/are withdrawn from consideration. 5) Claim(s) is/are allowed. 6) Claim(s) 8-11 is/are rejected. 7) Claim(s) is/are objected to. 8) Claim(s) are subject to restriction and/or election requirement. 						
Applicati	on Papers					
10)	The specification is objected to by the Examine The drawing(s) filed on <u>02 July 2006</u> is/are: a)[Applicant may not request that any objection to the Replacement drawing sheet(s) including the correction to the oath or declaration is objected to by the Ex	☐ accepted or b) ☐ objected to be drawing(s) be held in abeyance. See ion is required if the drawing(s) is obj	e 37 CFR 1.85(a). jected to. See 37 CFR 1.121(d).			
Priority u	ınder 35 U.S.C. § 119					
12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f). a) □ All b) □ Some * c) ☒ None of: 1. ☒ Certified copies of the priority documents have been received. 2. □ Certified copies of the priority documents have been received in Application No 3. □ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)). * See the attached detailed Office action for a list of the certified copies not received.						
2) Notic 3) Inform	e of References Cited (PTO-892) e of Draftsperson's Patent Drawing Review (PTO-948) nation Disclosure Statement(s) (PTO/SB/08) r No(s)/Mail Date <u>7/20/06</u> .	4) Interview Summary Paper No(s)/Mail Da 5) Notice of Informal P 6) Other: Examiner-an Christodoulatos.	ate			

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The disclosure is objected to because of the following informalities: At page 4, line 15, "necessitate" is misspelled. Based on page 9 lines 4-5 of PCT/JP2004/003874, ¹ it appears that page 10 line 17 should read "3.0 µm". At page 14 line 3, was "At the spot . . . " intended? Based on page 11 line 22 of PCT/JP2004/003874, it appears that page 14 line 10 should read "300 µm". At page 16 line 3, "time the volume. . . " should be "times the volume. . . " At page 17 line 5, "to enhance" should appear as "to enhance." At page 17 line 13, "achive" should appear as "achieve." Each instance of "Patent Document [x]" after paragraph [0018], where x is an element of {1,2,3,4,5}, should be replaced by the corresponding document number found at [0014] – [0018]. Appropriate correction is required.

Figure 2 should be designated by a legend such as --Prior Art-- because only that which is old is illustrated. See page 11 line 13. See MPEP § 608.02(g). Corrected drawings in compliance with 37 CFR 1.121(d) are required in reply to the Office action to avoid abandonment of the application. The replacement sheet(s) should be labeled "Replacement Sheet" in the page header (as per 37 CFR 1.84(c)) so as not to obstruct any portion of the drawing figures. If the changes are not accepted by the examiner, the applicant will be notified and informed of any required corrective action in the next Office action. The objection to the drawings will not be held in abeyance.

気液ミキサーを通過するに要する短時間内に、気泡径 0.5 5 ~ 3.0 μ m 程度、さらには 1 n m ~ 1,000 n m の超微 Application/Control Number: 10/586,487

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Claims 8 – 11 are rejected under 35 U.S.C. 112, second paragraph, as being indefinite for failing to particularly point out and distinctly claim the subject matter which applicant regards as the invention. Per claim 8, the phrase, "the decreasing rate of the concentration of the dissolved gas in the gasified liquid . . . is reduced " cannot be understood. It is unclear whether the "rate of concentration" of the gas into the liquid means the rate of dissolution of the gas from the gas phase to the liquid phase. Does applicant mean that the rate of dissolution of gas into the liquid phase decreases less quickly? If so, less quickly in comparison to which prior art method? Per claims 8, 10, and 11, it is unclear whether "the pressurized state" of claim 10 refers to the "pressurized condition" of claim 8 (this is a rejection based on lack of proper antecedent basis). Per claim 8, it is unclear what "instantaneously" means in this context. The dissolution of a gas phase chemical species, e.g., oxygen gas, into a water is necessarily a time dependent event, as any number of mass transfer textbooks or treatises will attest. If applicant means "rapidly" where "instantaneously" was used, then some objective standard by which "rapidly" may be distinguished from "slowly" would be required for the skilled artisan to adequately discern the metes and bounds of the claim. It is suggested that this basis for rejection be addressed through cancellation of the adverb, "instantaneously."

Claims 8 – 11 are rejected under 35 U.S.C. 112, first paragraph, as failing to comply with the enablement requirement. The claim(s) contains subject matter which was not described in the specification in such a way as to enable one skilled in the art to which it pertains, or with which it is most nearly connected, to make and/or use the

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invention. Although the application teaches how to dissolve oxygen into water, the application fails to teach how to do so "instantaneously." It is well known that dissolution of a gas such as oxygen into a liquid such as water is a mass transfer operation that is time-dependent albeit a relatively fast one under appropriate circumstances. See, for example, Hirota, et al, "Magnetic field effect on the kinetics of oxygen dissolution into water," Materials Transactions, 2000, vol. 41, no 8, pp. 976-980, or Chen, at Fig. 5 and related text.

Claims 8 – 11 are rejected under 35 U.S.C. 102(e) as being anticipated by Christodoulatos. USP 6752926 to Christodoulatos describes a process in which oxygen is instantaneously² dissolved in a recirculated wastewater line by means of a line atomizer, e.g., porous membrane 34 (col 8 line 21), under pressurization (Abstract), in advance and outside, i.e., upstream of and separate from, respectively, bioreactor reaction vessel 24. The reference implicitly admits that some oxygen bubbles are formed because it states that the formation of oxygen bubbles is minimized (Abstract) rather than prevented or eliminated. Alternatively, the skilled artisan would have understood that some air bubbles pass through the membrane in the embodiment in which the membrane is a porous one. Moreover, even in the case of a non-porous membrane, fine bubbles of oxygen would develop in the oxygenated water because there is a 2 – 4 psi pressure drop between the exit of the line atomizer 34 and the entrance to the bioreactor 24 (col 10 line 31). The skilled artisan would have understood that a portion of the oxygen that would be dissolved at the higher pressure

² "high rate of oxygen transfer" (Abstract)

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of the oxygenator would necessarily come out of solution in the form of fine bubbles as the pressure dropped those 2 – 4 psi. See for example col 3 line 55 at USP 4192742 to Bernard. Accordingly, the fluid that is admitted from the recirculation line to the lower right portion of the bioreactor reaction vessel 24 (as it appears in Fig. 3) would be a gasified solution introduced into the reaction vessel. The vessel is maintained in a pressurized condition. Per claim 9, the bioreactor includes packing for supporting biological organisms. Inherently, such packing meets the claim-recited functional limitations of increased habitat density, microorganism holding, and microorganism retention. Per claims 10 - 11, the degree of the pressurized state in the bioreactor, i.e., 20 psi, (Table 1) does not exceed the pressure at the outlet of the line atomizer, i.e., 22 – 24 psi.

Art cited of interest: USP 3928190 to Bébin describes a microorganism-supporting packed bed bioreactor receiving oxygenated water from an upstream outside line atomizer, but the bioreactor appears to be at ambient rather than elevated pressure. Other art cited of interest is found at: 7105092 5961830 5728305 5316682 5151187 4921604 4749527 4009105 4009099 3968035.

/Chester T. Barry/

Primary Examiner, Art Unit 1797

571-272-1152

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